

Ammonia Volatilization

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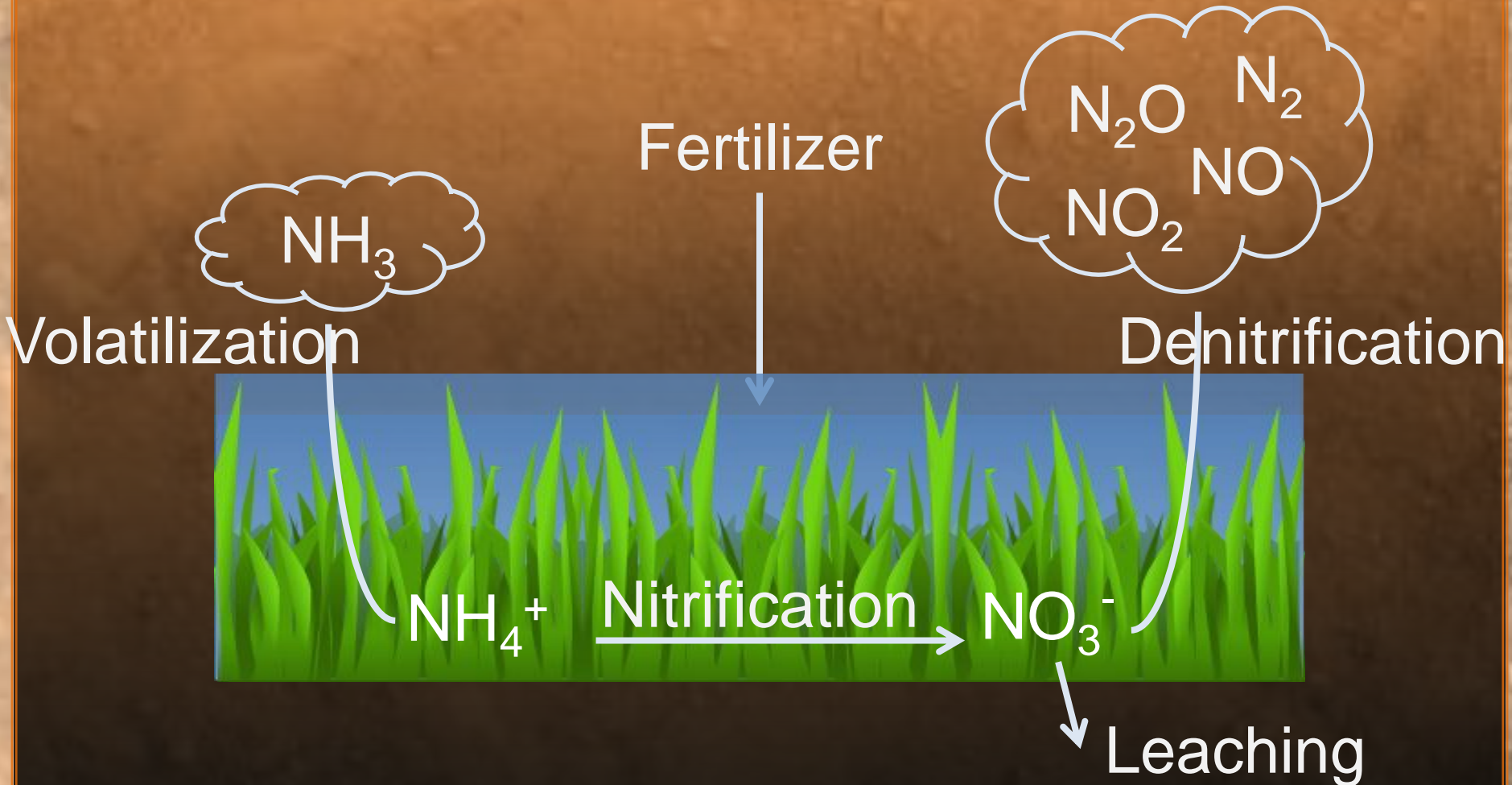
M.S. Soil Science

Oregon State University

Hermiston Agricultural Research and Extension Center (HAREC)

Hermiston, Oregon

Gas in the Nitrogen (N) Cycle



Ammonia Volatilization

- How it happens
- Factors increasing volatilization
 - How we measure it

The Problem

- Up to 50% of ammonium (NH_4^+) based and urea fertilizer lost as ammonia (NH_3)



- N fertilizer accounts for ~20% of NH_3 and ~70% of N_2O (anthropogenic, globally)

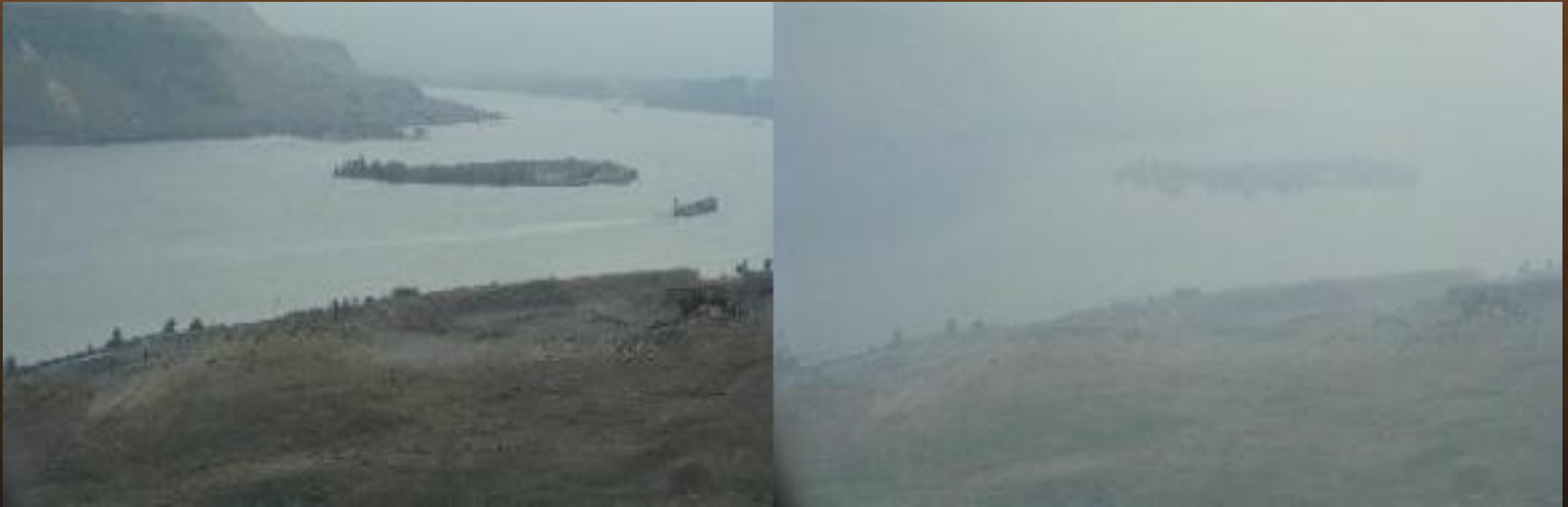
Why We Care About NH_3

- Reacts with SO_x and $\text{NO}_x \rightarrow$ aerosol particles



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Why We Care About NH_3

- Reacts with SO_x and NO_x → aerosol particles
- Unintended deposition to N sensitive ecosystems
- Nitrogen loss is economically bad

= \$\$\$



Factors Increasing Volatilization

Environmental

Moisture

High Soil Temperature

High wind speed

Low CEC

High Soil pH

Controllable... sometimes

Irrigation

Crop Residue

Fertilizer Type

Factors Increasing Volatilization

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Fertilizer Type

So many products...
Which ones work?

Choosing Fertilizer Additive to Limit Volatilization

- Urease inhibitor, NBPT
 - Active ingredient: (N-(n-butyl)thiophosphoric triamide)
 - In Agrotain[®] and others
 - >70% reduction of NH₃-N loss

Choosing Fertilizer Additive to Limit Volatilization

- Urease inhibitor, NBPT
- **Controlled/Slow release**
 - Degradable plastic coatings
 - Temperature
 - Moisture
 - Duration[®] & ESN[®]

Choosing Fertilizer Additive to Limit Volatilization

- Urease inhibitor, NBPT
- Controlled/Slow release
- **Ammonium sulfate (AS)**
 - All of N is in NH_4^+ form
 - >60% reduction of NH_3 -N loss
 - Results nearly as good as Agrotain[®]

Choosing Fertilizer Additive to Limit Volatilization

- Urease inhibitor, NBPT
- Controlled/Slow release
- Ammonium sulfate (AS)
- **Urea with AS**
 - Theory:
 - acidifying effects of AS nitrification
 - +
 - alkaline effects of urea hydrolysis
 - = no change in pH
 - Does it work?

Products

- (46-0-0) Urea
- (46-0-0) Urea + Agrotain[®]
- (21-0-0-24) AS
- (40-0-0-5) Urea + AS: separate
- (40-0-0-5) Urea + AS: fused
- (26-0-0-13) AS Nitrate (ASN)
- (39-0-0-12) Sulfur-coated urea (SCU)



Study Goals

- Find methods and products that limit NH_3 volatilization
- Estimate how much NH_3 is *actually* lost

Methods

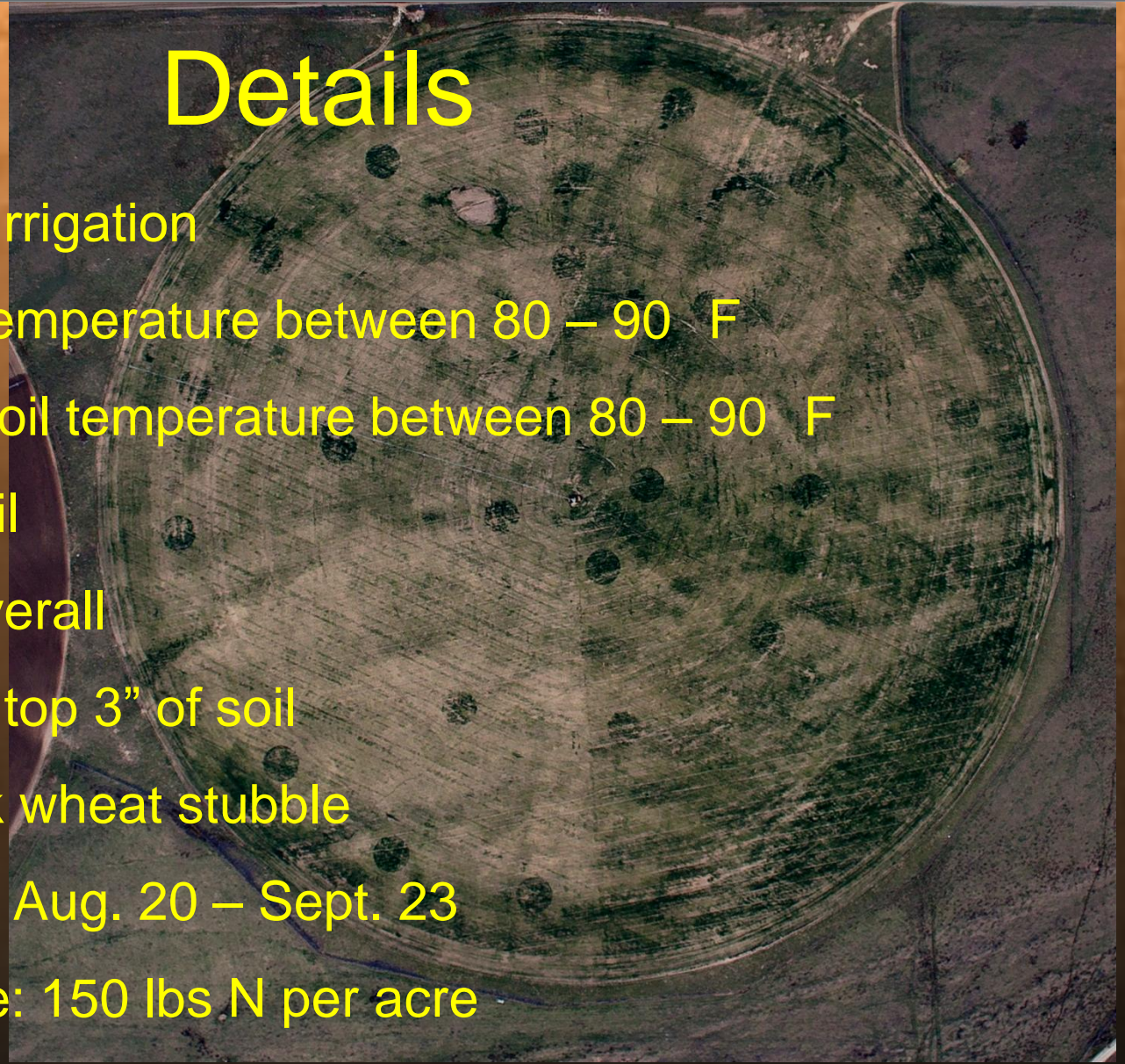
- 10' poles with wind vane
- Passive = no power, no influence on environment
- Tubes coated with oxalic acid: traps $\text{NH}_3 \rightarrow \text{NH}_4^+$
- Lab analysis of $[\text{NH}_4^+]$ translates into actual NH_3 loss estimate



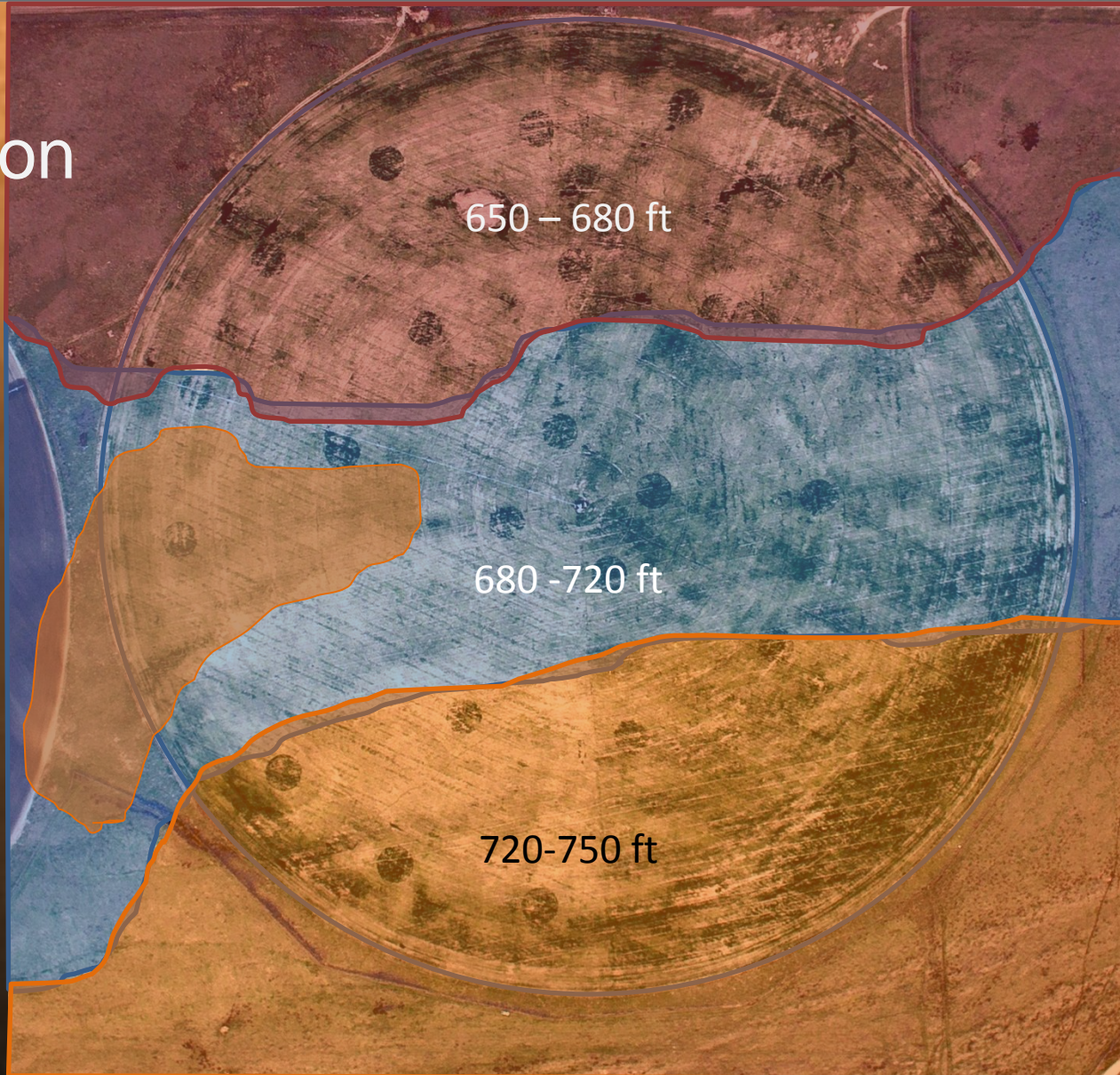


Details

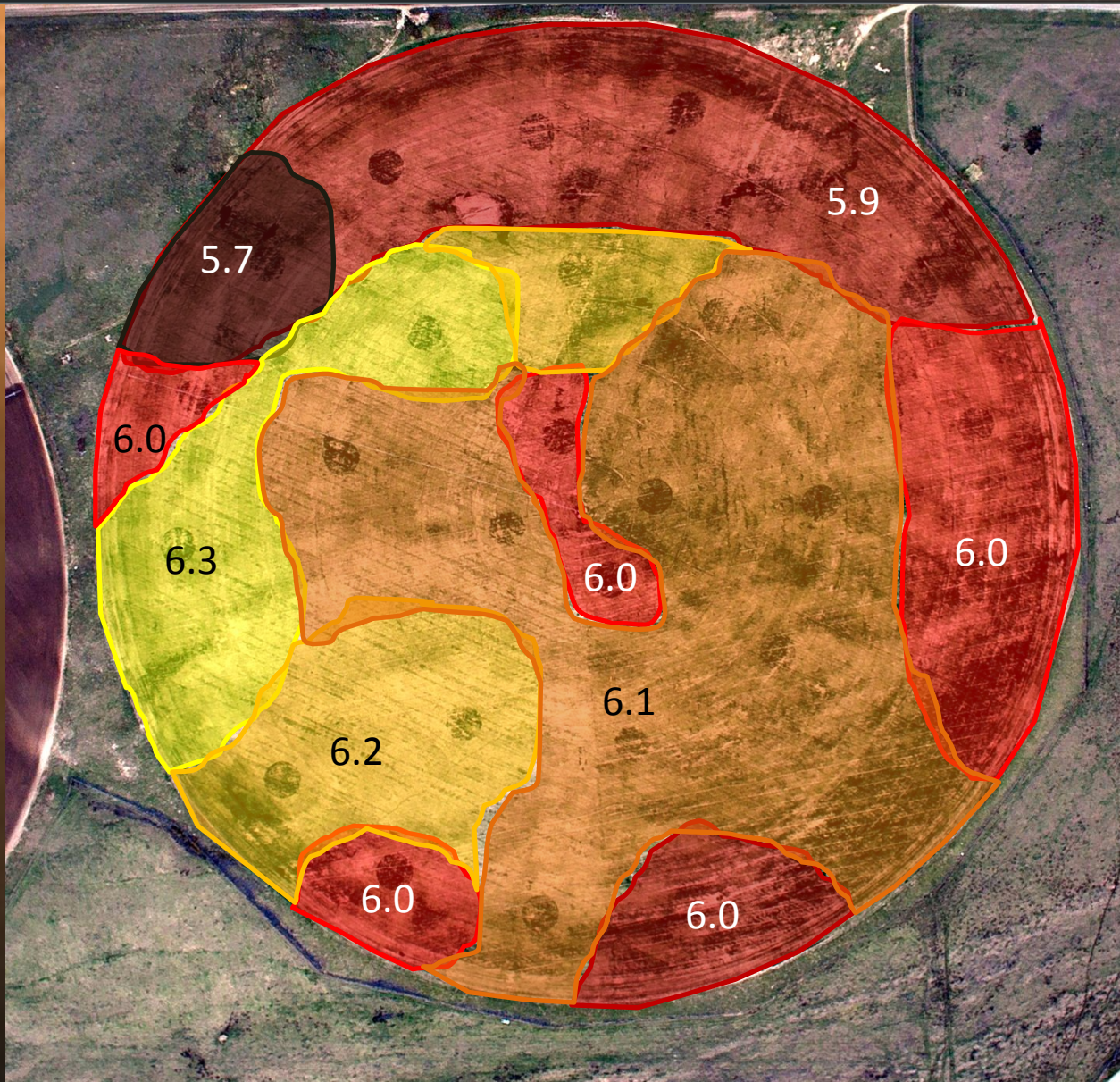
- Pre-applied 2" irrigation
- Average high temperature between 80 – 90 F
- Average high soil temperature between 80 – 90 F
- Sandy loam soil
- pH 7.3 – 7.9 overall
- pH 5.8 – 6.3 in top 3" of soil
- Applied to thick wheat stubble
- Study duration: Aug. 20 – Sept. 23
- Application rate: 150 lbs N per acre



Elevation

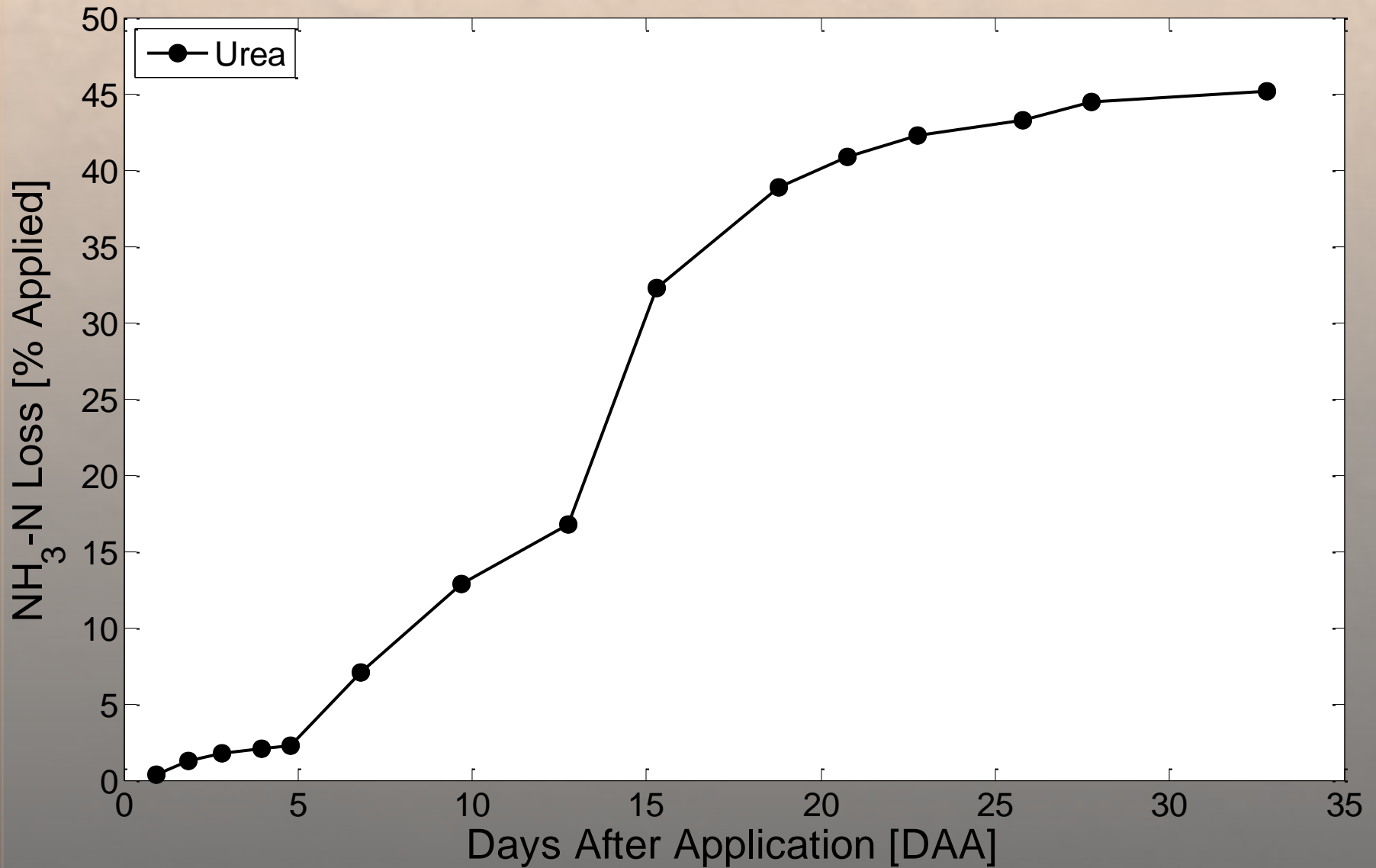


pH

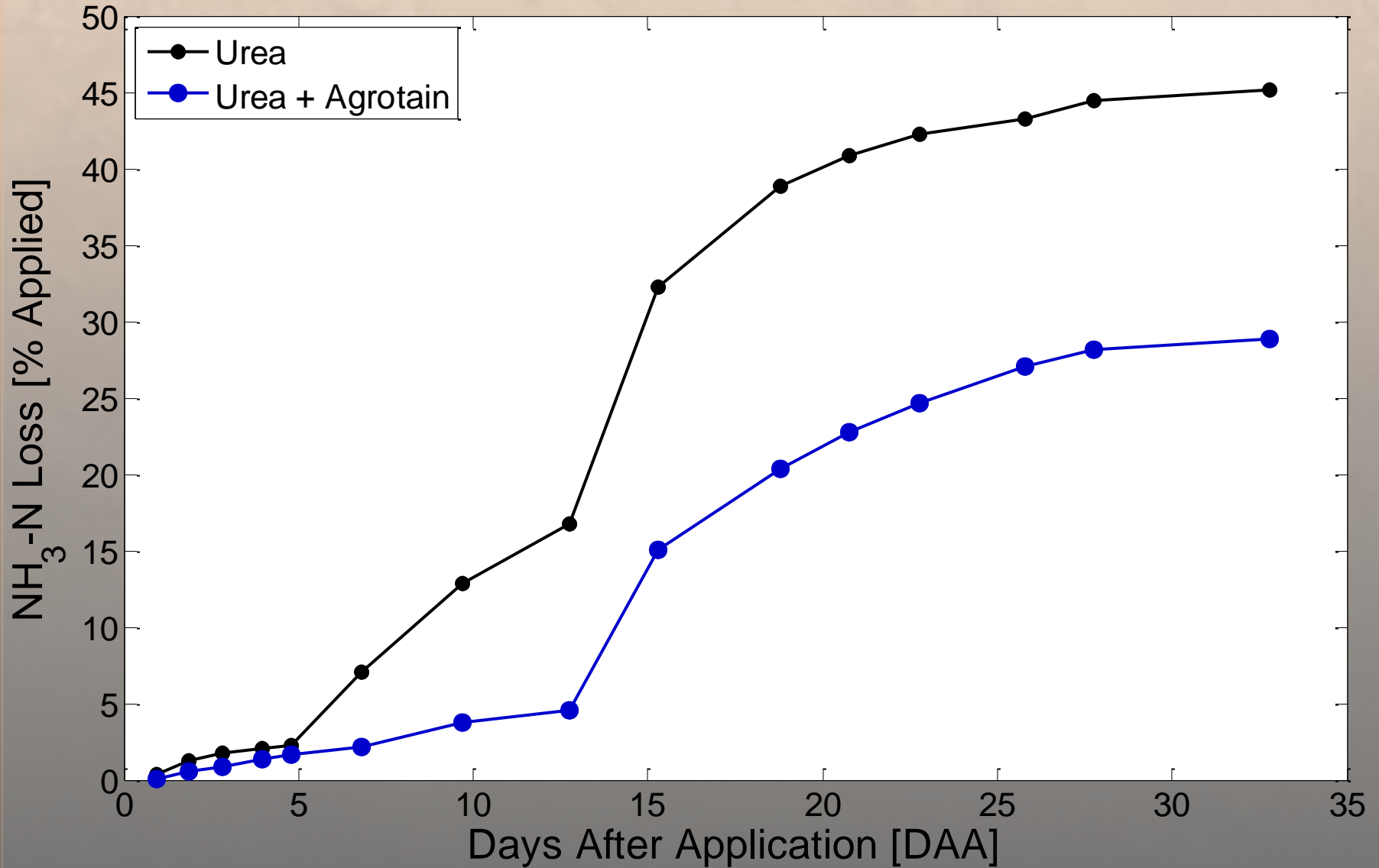


Results

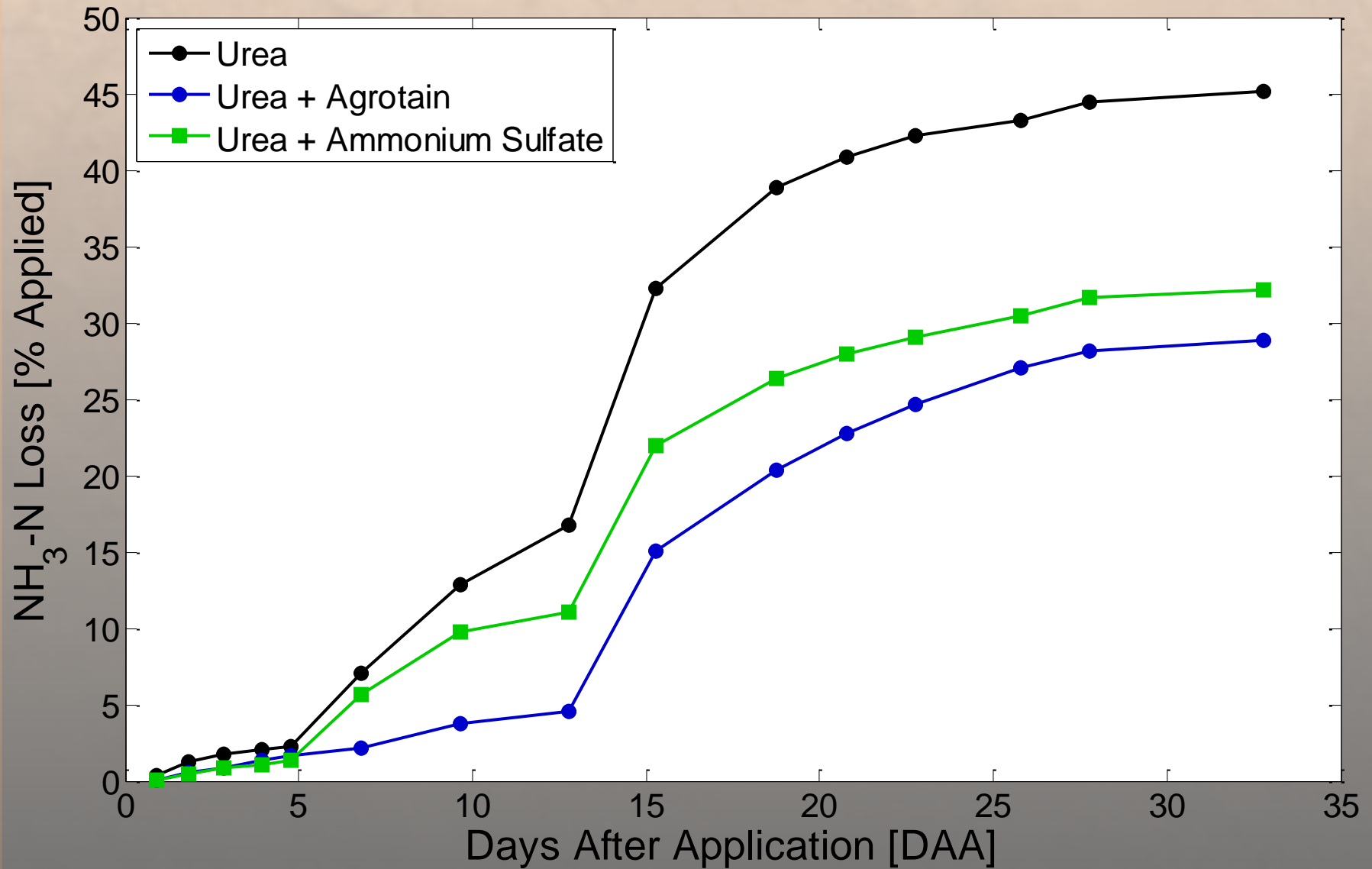
Cumulative NH_3 -N Loss by Treatment



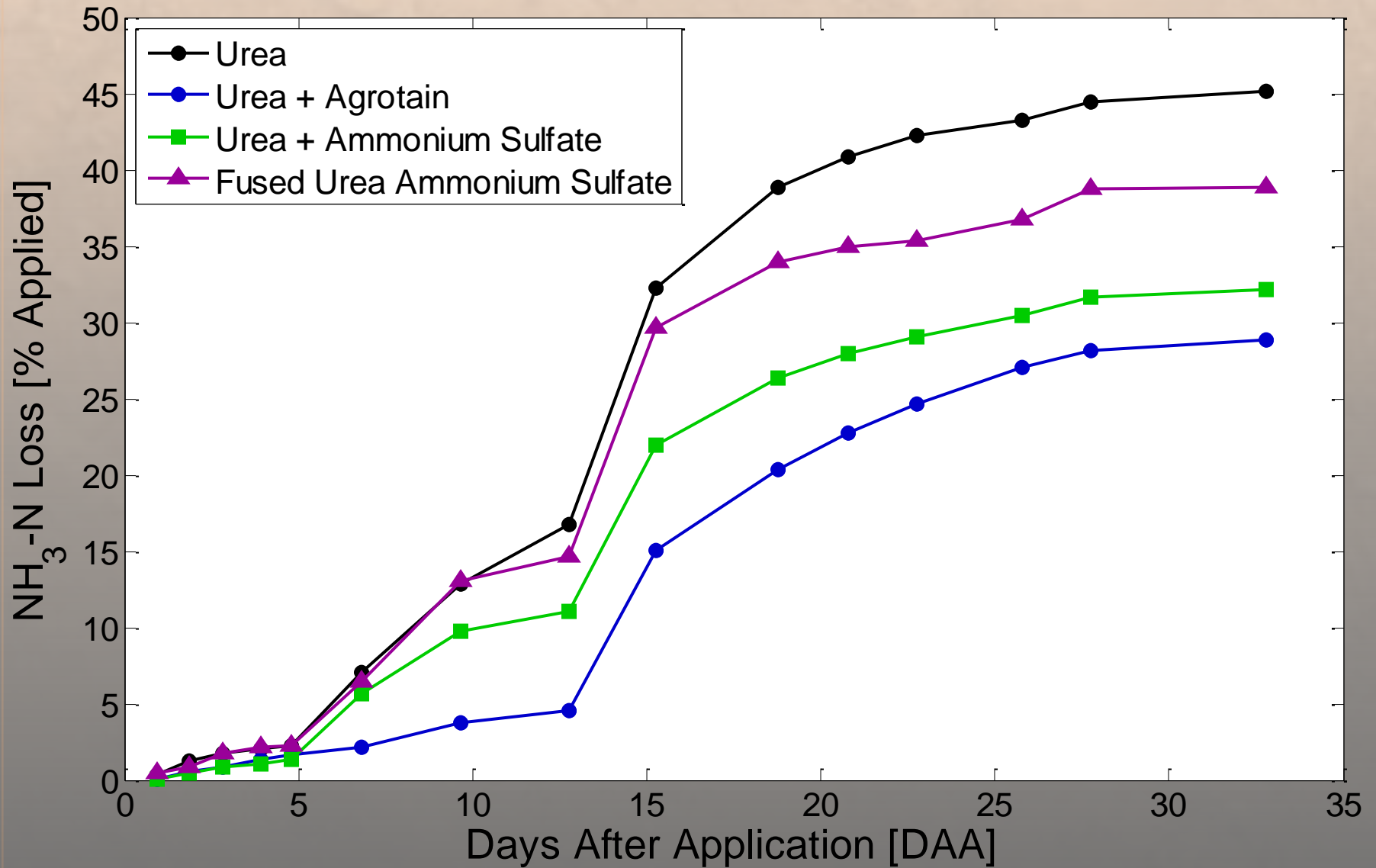
Cumulative NH_3 -N Loss by Treatment



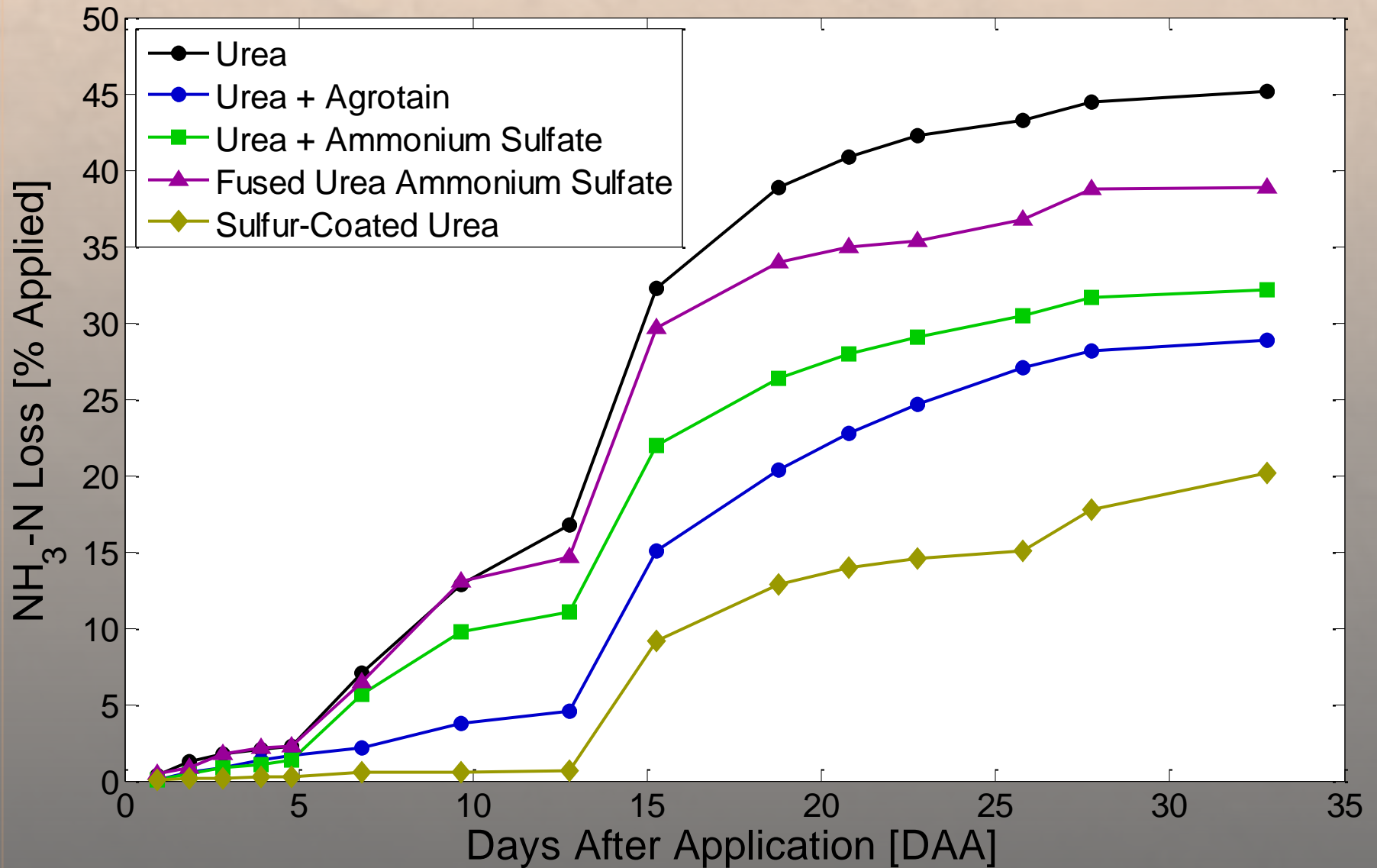
Cumulative NH_3 -N Loss by Treatment



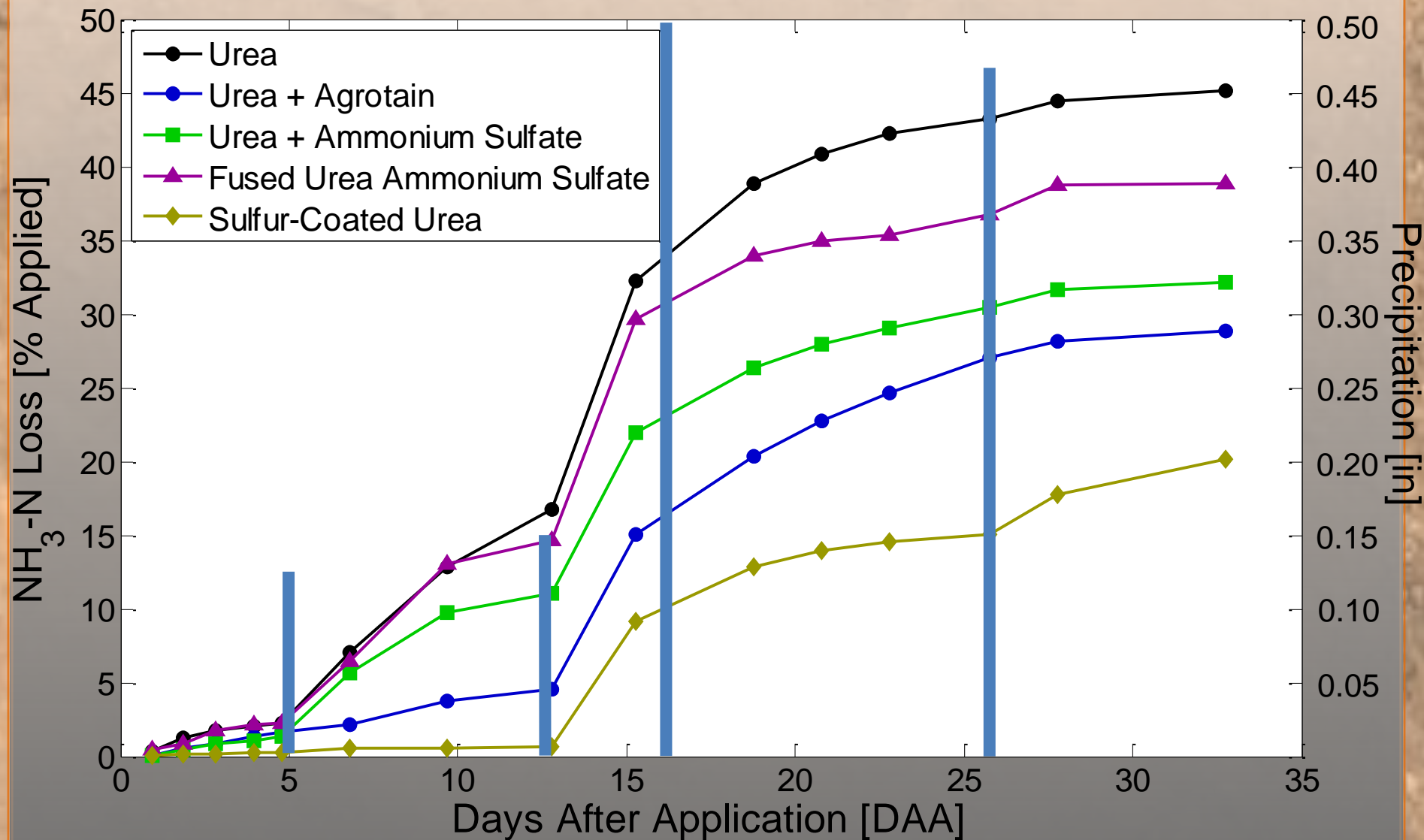
Cumulative NH_3 -N Loss by Treatment



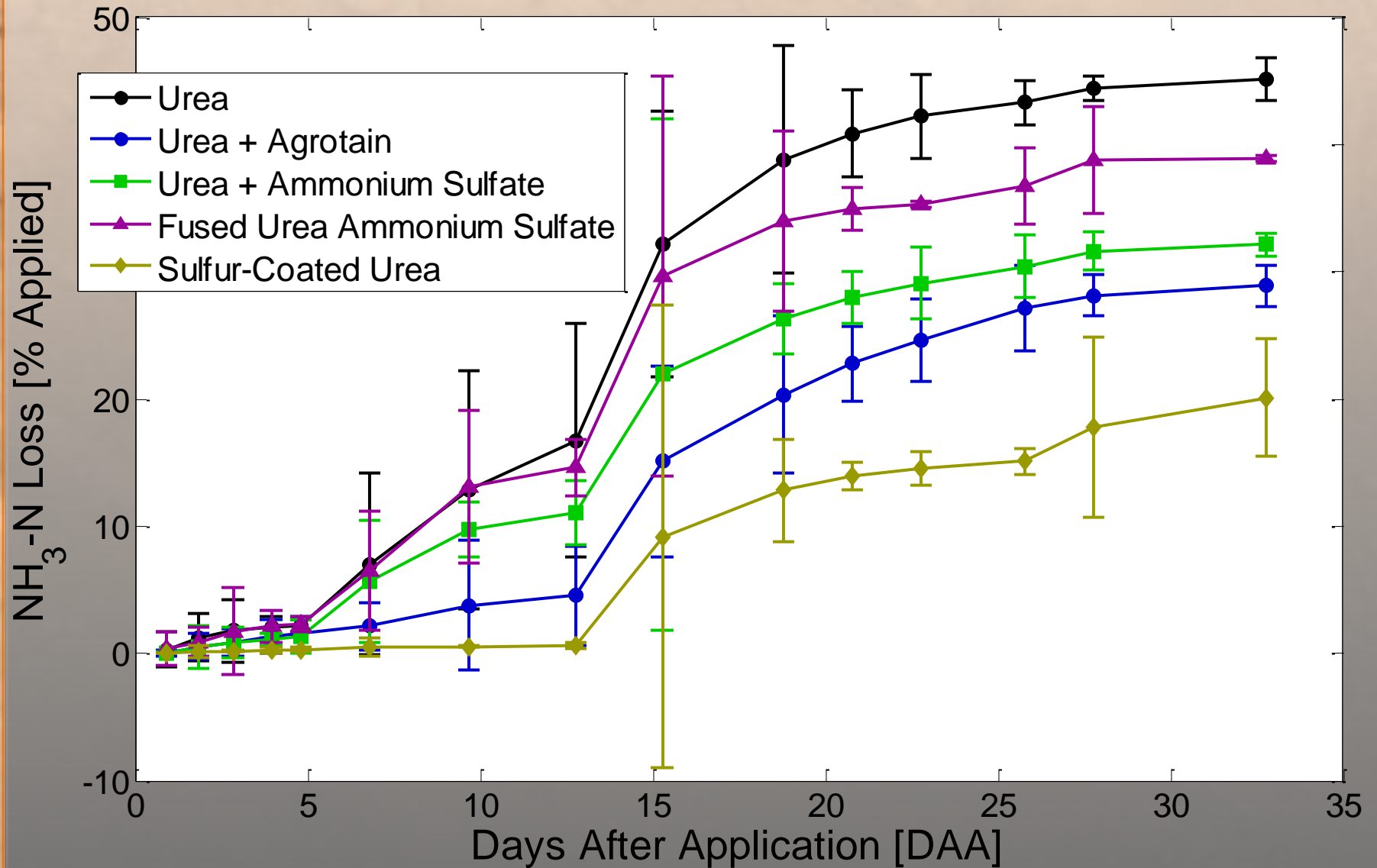
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Cumulative NH_3 -N Loss by Treatment



Conclusions

- In a localized race between the hydrolysis reaction of urea and nitrification of AS, urea hydrolysis wins

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- In a localized race between the hydrolysis reaction of urea and nitrification of AS, urea hydrolysis wins
- Hydrolysis occurs at a faster initial rate
- AS provides extra NH_4^+ to volatilize until urea hydrolysis slows down.

Ways to Reduce NH_3 Volatilization

- Incorporate fertilizer
 - Irrigation
 - Tillage
- Apply in favorable environmental conditions
 - Not 100° outside or after a mild rain
- Choose fertilizer type or additive that reduces loss depending on environmental conditions and loss mechanism
 - Inhibitors, controlled/slow release, NH_4^+ , NO_3^- , or urea based fertilizer

Acknowledgements

- Yara International
- Koch Industries, Inc.
- J.R. Simplot Company
- Two Rivers Terminal, LLC
- The growers that make it possible
- My major professor, mentors from campus
- HAREC

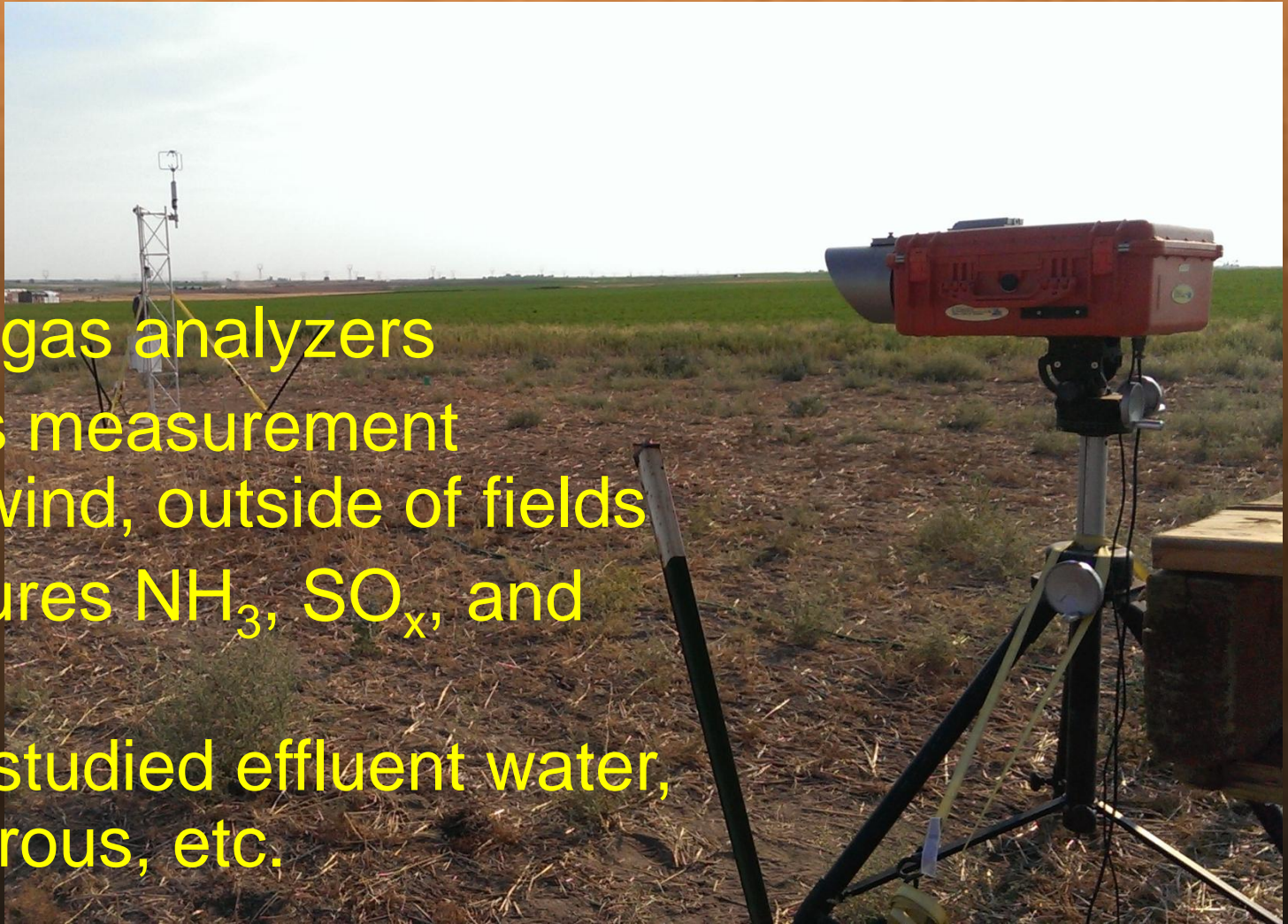
Concurrent Research

- Fertigation studies using our mini-center pivots: 61 ft, irrigates ~1/5 acre



Concurrent Research

- Laser gas analyzers
- Allows measurement downwind, outside of fields
- Measures NH_3 , SO_x , and NO_x
- Have studied effluent water, anhydrous, etc.



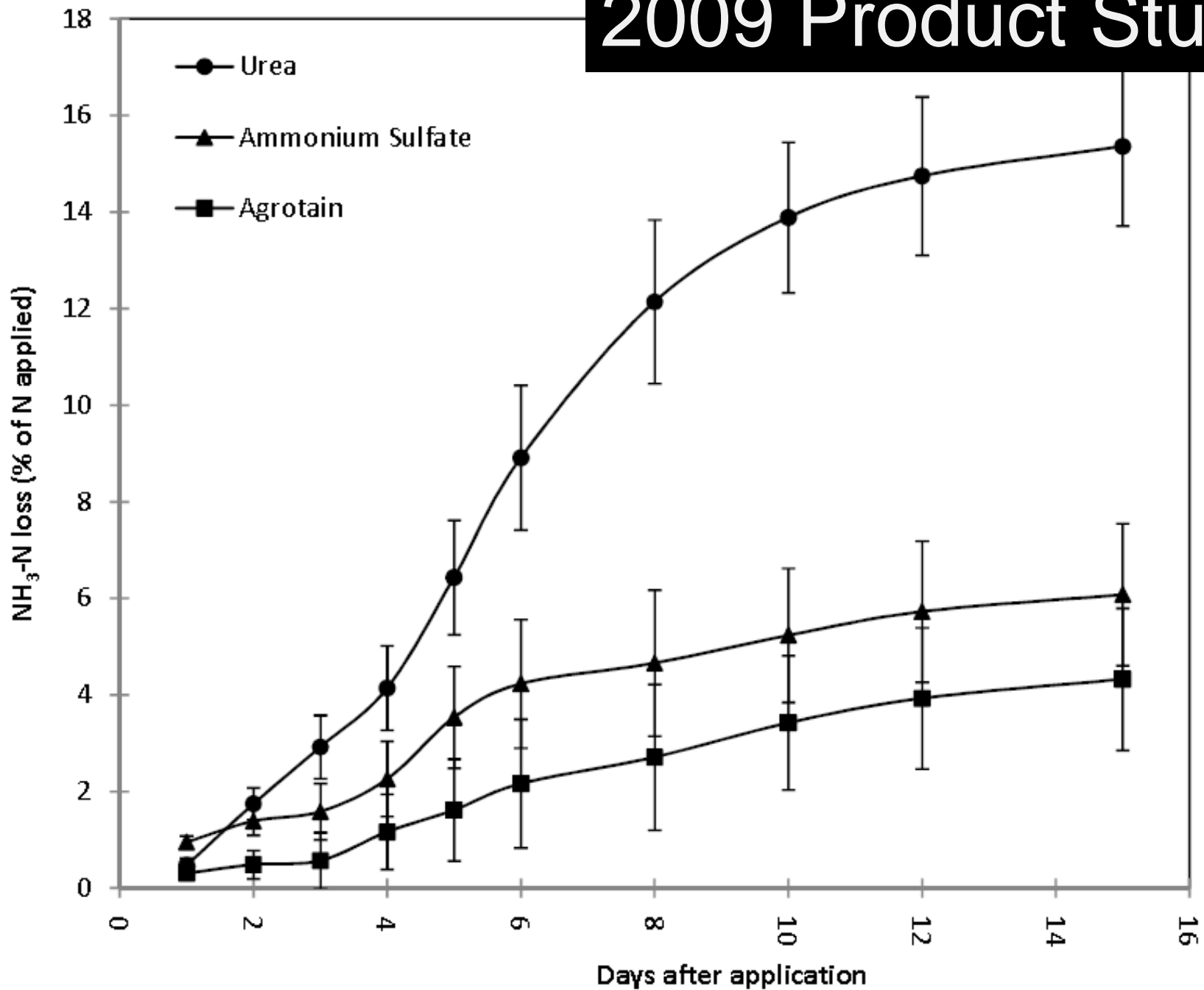
A microscopic image of a plant cell, likely a cross-section of a stem or root, showing distinct layers of cells. The image is stained with red and blue dyes, highlighting various cellular structures. The central part of the cell is filled with a dense network of red-stained fibers, possibly cellulose or lignin. The outer layers show more organized cell walls and structures. The overall appearance is that of a complex, multi-layered biological structure.

Questions?

Thank you!

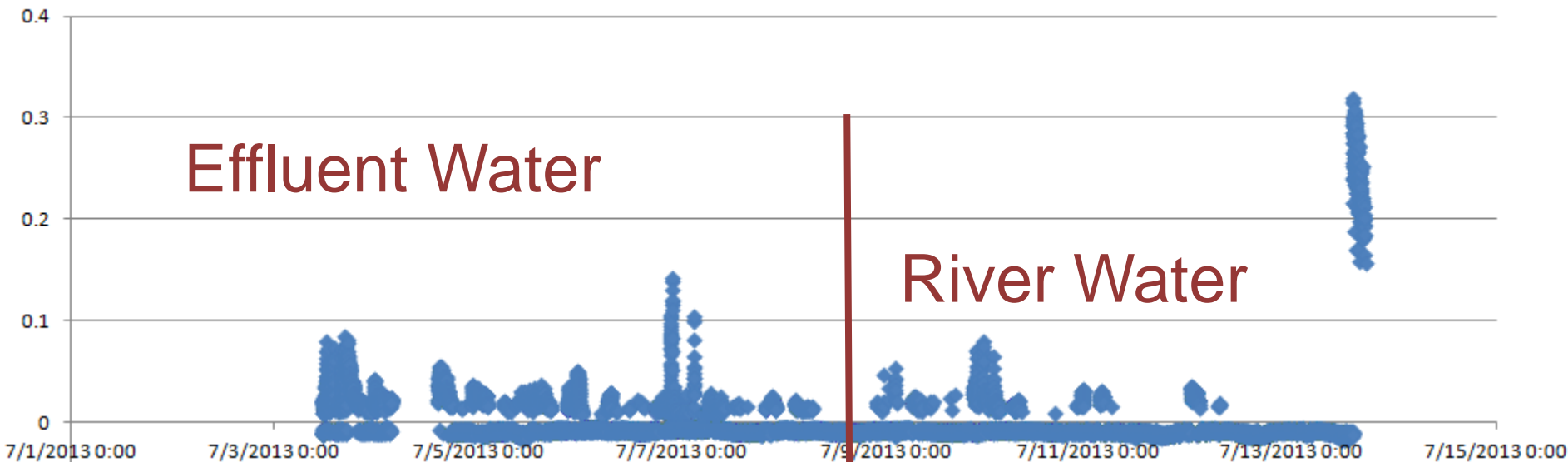
Appendix Slides

2009 Product Study



Effluent Results

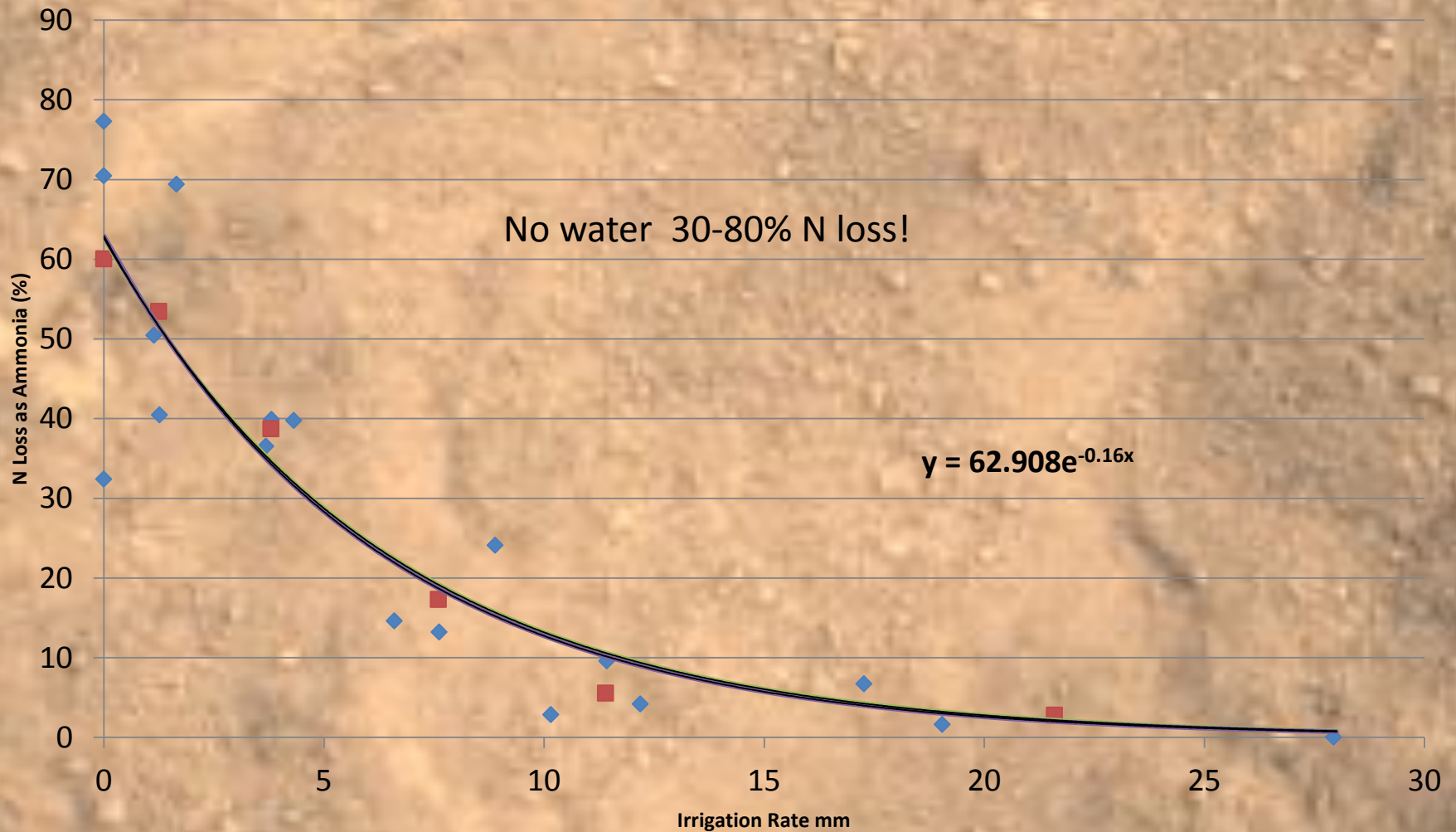
Ammonia Volatilization from Effluent Water



- Effluent estimated to contain 100 ppm TKN
- Need to be cleaned up still
- Large amounts of data contains some noise

Effect of irrigation

Irrigation vs. Loss as ammonia



Agrotain vs. Urea

